Record Nr. UNISA996211962403316 Formation of bonds to C, Si, Ge, Sn, Pb . Part 1 [[electronic resource] /] **Titolo** / founding editor, J.J. Zuckerman; editor, A.P. Hagen Pubbl/distr/stampa New York, N.Y., : VCH Publishers, 1991 **ISBN** 1-282-30813-0 9786612308130 0-470-14523-4 0-470-14544-7 Descrizione fisica 1 online resource (633 p.) Collana Inorganic reactions and methods::9 ZuckermanJ. J <1936-1987.> (Jerold J.) Altri autori (Persone) HagenA. P Disciplina 541.3/9 541.39 Soggetti Chemical kinetics - Effect of temperature on Inorganic compounds - Synthesis Lingua di pubblicazione Inglese **Formato** Materiale a stampa Livello bibliografico Monografia Description based upon print version of record. Note generali Nota di bibliografia Includes bibliographical references and indexes. Nota di contenuto Inorganic Reactions and Methods; Contents; How to use this book; Preface to the Series; Editorial Consultants to the Series; Contributors to Volume 9; Formation of the Bonds to the Group-IVB (C, Si, Ge, Sn, Pb) Elements; Introduction; Formation of the Group-IVB (C, Si, Ge, Sn, Pb)-Group-IVB (C, Si, Ge, Sn, Pb) Element Bond; Introduction; Formation of the Carbon-Carbon Bond; Formation of the Si-Si Bond; in Elemental Silicon; from Oxides.; by Other Methods.; Formation of High-Purity Silicon: Electrochemical Formation of Compounds with Si-Si Bonds (Including Elemental Silicon) from Metal Silicidesby Solvolysis in Aqueous or Liquid Ammonia Solutions.; Formation of Oligosilanes; Formation of Polymeric Compounds; from Silicon Hydrides and Organosilicon Hydrides; by the Action of a Silent Electric Discharge.; by Direct Photolysis.; by Sensitized Photolysis.; by Catalyzed Reactions.; by Reaction with SilvImetallic Compounds.; from Silicon Halides and Organosilicon Halides; by Electrochemical Reduction.; by Halide Elimination with

Active Metals.; by Reaction with SilvImetallics.; by Catalyzed

Disproportionation.; by Reaction with Organomagnesium Halide Reagents.

by the Action of Silent Electric Discharge.by Mercury-Photosensitized Photolyses.; from Bissilylmercury Compounds; by Thermolysis.; by Photolysis.; from Organosilanes and Silicon Halides by Hydrogenolysis.; from Silylenes; by Oligomerization.; by Insertions into Bonds of Silicon to Hydrogen, Oxygen and Silicon.; by Addition to Si = C.; in the Direct Reaction of Methyl Chloride with Silicon-Copper.: Formation of the Germanium-Germanium Bond; in Elemental Germanium; from Oxides.: from Sulfides.; by Other Syntheses.; Formation of High-Purity Germanium.; from Organogermanium Hydrides by Reaction with Diorganomercury Compounds in the Presence of UV Radiation.by Hydrogermolysis Reaction.; by Germanium Hydride Decomposition.: from Ge(IV) Halides and Organogermanium(IV) Halides; by the Action of a Microwave Discharge of Ge(IV) Halides.; by Electrochemical Reduction.; by Halide Elimination with Active Metals.; by Reaction with Germyl-Metal Reagents.; by Reaction with Organometallic Reagents.: from Germanium(II) Halides: by Reactions with Germyl-Metal Reagents.; by Reaction with Organometallic Reagents.; from Germyl Compounds of Cadmium, Mercury, Thallium, Antimony and Bismuth by Thermolysis or Photolysis.from Germylenes; by Oligomerization of Germylenes.; by Insertions into Bonds of Germanium to Hydrogen, Halogen, Carbon, Oxygen, Sulfur, Nitrogen, Phosphorus and Germanium.: The Formation of the Tin-Tin Bond: in Elemental Tin: from Oxides.; from Sulfides.; by Other Syntheses.; Allotropy of Tin; from Organotin Hydrides; by Catalytic Hydrogen Elimination.; by Reaction with Organotin Halogen, Pseudohalogen, Chalcogen and Pnictogen Compounds.; by Reaction with Organometallic Compounds.; by Reaction with Reducible Organic Compounds.; from Organotin Halides

Sommario/riassunto

For the first time the discipline of modern inorganic chemistry has been systematized according to a plan constructed by a council of editorial advisors and consultants, among them three Nobel laureates (E.O. Fischer, H. Taube and G. Wilkinson).Rather than producing a collection of unrelated review articles, the series creates a framework which reflects the creative potential of this scientific discipline. Thus, it stimulates future development by identifying areas which are fruitful for further research. The work is indexed in a unique way by a structured system which maximize

by Electrochemical Reduction.