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| 1. Record Nr. | UNISA990001036130203316 |
| Titolo | International society Stanford law school : proceedings |
| Pubbl/distr/stampa | Stanford : Stanford law school, 1966-1967 |
| Descrizione fisica | 2 v. ; 23 cm |
| Disciplina | 341.06 |
| Soggetti | Diritto internazionale -- Periodici |
| Lingua di pubblicazione | Inglese |
| Formato | Materiale a stampa |
| Livello bibliografico | Periodico |
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| 2. Record Nr. | UNIORUON00300845 |
| Autore | SILONE, Ignazio |
| Titolo | Fontamara / Ignazio Silone ; versiun romontscha e introducziun da Ursicin G.G. Derungs |
| Pubbl/distr/stampa | [Vella], : U.G.G. Derungs, stampa 2003 |
| Descrizione fisica | 237 p. ; 19 cm. |
| Disciplina | 853.91 |
| Lingua di pubblicazione | Francese |
| Formato | Materiale a stampa |
| Livello bibliografico | Monografia |

3. Record Nr.	UNINA9911019249303321
Autore	Kromidas Stavros
Titolo	More practical problem solving in HPLC // Stavros Kromidas ; with contributions by Friedrich Mandel, Jurgen Maier-Rosenkranz and Hans-Joachim Kuss
Pubbl/distr/stampa	Weinheim ; ; [Great Britain], : Wiley-VCH, c2005
ISBN	9786611239268 9781281239266 1281239267 9783527612307 3527612300 9783527612314 3527612319
Descrizione fisica	1 online resource (312 p.)
Disciplina	543.84
Soggetti	High performance liquid chromatography Phase partition
Lingua di pubblicazione	Inglese
Formato	Materiale a stampa
Livello bibliografico	Monografia
Note generali	Includes index.
Nota di contenuto	More Practical Problem Solving in HPLC; Foreword; Contents; Preface; The Structure of the Book; Part 1 (general section); Part 2 (specific questions); In Lieu of an Introduction; Chromatography - and more - Crossword; Across; Down; An HPLC-Quiz; An HPLC Tale; The Tale of Peaky and Chromy; 1 HPLC Tips; 1.1 Stationary Phases and Columns; Tip No. 01 "It improves with age" is a rule that applies to port and sometimes to red wine, but how about your C(18) column?; Tip No. 02 Optimization via column parameters - what works best? Tip No. 03 Can selectivity always be put down to chemical interactions with the stationary phase? Tip No. 04 A matter of perspective . . . Or: Selectivity and peak symmetry of basic compounds using reversed-phase packing materials; Tip No. 05 Separation of isomers; Tip No. 06 When should I use a "polar" C(18) phase?; Tip No. 07 Are polar RP-C(18) phases more suitable for the separation of polar analytes than non-polar phases?; Tip No. 08 What about non-encapped phases -

are they a thing of the past?; Tip No. 09 How can I separate acids using RP C(18)?

Tip No. 10 The nitrile phase - some like it polar; Tip No. 11 The selectivity of RP columns; 1.2 Buffers, pH Value; Tip No. 12 Does it always have to be potassium phosphate?; Tip No. 13 UV cut-off of buffer solutions; Tip No. 14 Sources of errors when using buffers; Tip No. 15 The drawbacks of using buffers; Tip No. 16 Why is the pH value so important, and what does it do?; Tip No. 17 Why does the pH value shift even though I am using the correct buffer and the buffer capacity is sufficient?; Tip No. 18 Changes to the pH value in the eluent: the extent of the shift and the reasons behind it

Tip No. 19 An unintentional pH shift and its consequences; Tip No. 20 RP separations in the alkaline medium; Tip No. 21 Separation of basic and acidic compounds contained in the same sample; 1.3 Optimization, Peak Homogeneity; Tip No. 22 The peaks appear too soon - what can be done?; Tip No. 23 What can I do if the peaks elute late?; Tip No. 24 Quick optimization of an existing gradient method; Tip No. 25 Increasing efficiency - often the fast track to success; Tip No. 26 Additives to the eluent; Tip No. 27 Separating the unknown - where shall I begin?

Tip No. 28 Separation of an unknown sample using a reversed-phase C (18) column - how do I go about it?; Tip No. 29 Developing an RP

separation - the two-day-method; Part 1: Choice of column and eluent;

Tip No. 30 Developing an RP separation - the two-day method; Part 2:

Fine-tuning of the separation; Tip No. 31 Quick check on peak

homogeneity; Tip No. 32 Quick check on peak homogeneity; Tip No. 33

Tied to a standard operating procedure - how can a bad separation be

improved further?; Tip No. 34 More elaborate measures to check peak

homogeneity; Tip No. 35 First easily digestible tip

Tip No. 36 Second easily digestible tip

Sommario/riassunto

A unique approach to solving HPLC problems. Everyone who bought ""Problem Solving in HPLC"" by Stavros Kromidas will equally benefit from nearly 100 new practical examples for optimization, trouble-shooting, and instrument performance given in this sequel. The author provides- guidance for selecting and evaluating methods, instruments and columns,- practical help with everyday trouble-shooting,- advice for optimizing separations, always explaining the reason why. In each case the problem, the solution and the conclusions are presented over a maximum of 4 pages, and in add
