1. Record Nr. UNINA9910454918403321

Autore	Norris Christopher
Titolo	The contest of faculties [[electronic resource]] : philosophy and theory after deconstruction / / Christopher Norris
Pubbl/distr/stampa	Abingdon, Oxon, : Routledge, 2010
ISBN	1-136-99901-9 1-282-97404-1 9786612974045 0-203-85564-7
Descrizione fisica	1 online resource (170 p.)
Collana	Routledge revivals
Disciplina	801.95
Soggetti	Deconstruction Narration (Rhetoric) Literature - Philosophy Electronic books.
Lingua di pubblicazione	Inglese
Formato	Materiale a stampa
Livello bibliografico	Monografia
Note generali	Description based upon print version of record.
Nota di bibliografia	Includes bibliographical references and index.
Nota di contenuto	Book Cover; Title01; Copyright01; Title02; Copyright02; Contents; Acknowledgements; Introduction: philosophy, theory and the 'contest of faculties'; 1 Narrative theory or theory-as-narrative: the politics of 'post-modern' reason; 2 Sense, reference and logic: a critique of post-structuralist theory; 3 Some versions of rhetoric: Empson and de Man; 4 Transcendent fictions: imaginary discourse in Descartes and Husserl; 5 Aesthetics and politics: reading Roger Scruton; 6 Philosophy as a kind of narrative: Rorty on post-modern liberal culture; 7 Suspended sentences: textual theory and the Law 8 On not going relativist (where it counts): deconstruction and 'Convention T'9 Conclusion; Notes; Index
Sommario/riassunto	This Routledge Revival, first published in 1985, gives detailed attention to the bearing of literary theory on questions of truth, meaning and reference. On the one hand, deconstruction brings a vigilant awareness of the figural and narrative tropes that make up the discourse of philosophic reason. On the other it insists that argumentative rigour cannot be divorced from the kind of close reading that has come to

		characterize literary theory in its more advanced or speculative forms. This present-day 'contest of faculties' has large implications for philosophers and critics, many o

2.	Record Nr.	UNINA9910134909803321
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	Soggetti	Free trade - Europe Europe Commercial policy Periodicals
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	Formato	Materiale a stampa
	Livello bibliografico	Periodico
	Note generali	Title from journal home page (publisher Web site, viewed July 23, 2004).

3.	Record Nr.	UNINA9911019494003321
	Titolo	The experimental determination of solubilities / / edited by G.T. Hefter and R.P.T. Tomkins
	Pubbl/distr/stampa	Chichester, West Sussex, England ; ; Hoboken, NJ, : J. Wiley & Sons, c2003
	ISBN	9786610271825 9780470867839 9780471497080
	Descrizione fisica	1 online resource (659 p.)
	Collana	Wiley series in solution chemistry ; ; v. 6
	Altri autori (Persone)	HefterG. T TomkinsR. P. T (Reginald P. T.)
	Disciplina	541.3/42
	Soggetti	Solubility
	Lingua di pubblicazione	Inglese
	Formato	Materiale a stampa
	Livello bibliografico	Monografia
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Nota di contenuto

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7. Concluding Remarks8. References; Appendix A: Some Useful Thermodynamic Concepts and Relations; Appendix B: Numerical and Statistical Procedures in Constructing Fitting Equations; Chapter 1.2 Kinetics and Mechanisms of Crystal Growth and Dissolution; 1. Introduction; 2. Fundamental Concepts; 3. Mechanisms and Rate Expressions for Dissolution and Growth; 4. Comparison of Dissolution and Precipitation Kinetics; 5. Kinetics of Approach to Equilibrium; 6. Summary of Rate-determining Mechanisms; 7. Acknowledgement; 8. References; 2 GASES; Chapter 2.1 Solubility of Gases in Liquids
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Chapter 4.3 Solubility of Salt-Water Systems at High Temperatures and Pressures

Sommario/riassunto

This book covers the most useful experimental methods for all types of solubility measurements. The importance of solubility phenomena has been long recognized throughout science. For example, in medicine, the solubility of gases in liquids forms the basis of life itself; in the environment, solubility phenomena influence the weathering of rocks, the creation of soils, the composition of natural water bodies and the behaviour and fate of many chemicals. However, until now, no systematic critical presentation of the methods for obtaining solubilities has been given.