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Concluding Remarks; References

3. Solution Combustion Synthesis of Oxide Materials 3.1 Introduction; 3.2 Solution Combustion Synthesis (SCS); 3.2.1 Synthesis of Alumina; 3.2.2 Mechanism of Aluminum Nitrate - Urea Combustion Reaction; 3.2.3 Thermodynamic Calculation; 3.3 Role of Fuels; 3.4 A Recipe for the Synthesis of Various Classes of Oxides; 3.4.1 Recipe for Nanomaterials; 3.5 Salient Features of Solution Combustion Method; References; 4. Alumina and Related Oxide Materials; 4.1 Introduction; 4.2 Alumina and Related Oxide Materials; 4.3 -Alumina; 4.4 Metal Aluminates (MAl_2O_4); 4.5 Rare Earth Orthoaluminates (LnAlO_3) 4.6 Garnets 4.7 Aluminum Borate; 4.8 Tialite ($-\text{Al}_2\text{TiO}_5$); 4.9 Aluminum Phosphate; 4.10 Alumina Composites; 4.10.1 $\text{Al}_2\text{O}_3 \cdot \text{SiO}_2$ System: Mullite; 4.10.2 $\text{Al}_2\text{O}_3 \cdot \text{SiO}_2$ System: Cordierite; 4.10.3 $\text{Al}_2\text{O}_3 \cdot \text{Si}_3\text{N}_4$ System: SiAlON ; 4.11 Alumina Nanocomposites; 4.11.1 Nanocatalysts, Dispersion of Nano-metals (Ag, Au, Pd, and Pt) in Al_2O_3 ; 4.12 Nanopigments; 4.12.1 Cobalt-Based Blue Alumina and Aluminates; 4.12.2 Chromium-Doped Pink Alumina ($\text{Cr}^{3+}/\text{Al}_2\text{O}_3$): Ruby; 4.12.3 Chromium-Doped Aluminates and Orthoaluminates ($\text{Cr}^{3+}/\text{MAl}_2\text{O}_4$ ($\text{M} = \text{Mg} \ \& \ \text{Zn}$)) and LaAlO_3 ; 4.13 Nanophosphors 4.13.1 Phosphor Materials (Luminescence in Aluminum Oxide Hosts) 4.14 Concluding Remarks; References; 5. Nano-Ceria and Metal-Ion-Substituted Ceria; 5.1 Introduction; 5.2 Synthesis and Properties of Nano-Ceria; 5.3 Synthesis of Metal-Ion-Substituted Ceria; 5.4 Characterization of Metal-Ion-Substituted Ceria; 5.5 Oxygen Storage Materials; 5.6 Metal-Ion-Substituted Ceria as Nanocatalysts; 5.6.1 $\text{Ce}_{1-x}\text{PdxO}_2$ as a Three-Way Catalyst; 5.6.2 $\text{Ce}_{1-x}\text{Pt}_x\text{O}_2$; 5.6.3 $\text{Ce}_{1-x}\text{Rh}_x\text{O}_2$; 5.6.4 Bimetal Ionic Catalysts ($\text{Ce}_{1-x}\text{Pt}_x/2\text{Rh}_x/2\text{O}_2$); 5.7 Concluding Remarks; References; 6. Nanocrystalline Fe_2O_3 and Ferrites 6.1 Magnetic Materials

Sommario/riassunto

Nano-oxide materials lend themselves to applications in a wide variety of emerging technological fields such as microelectronics, catalysts, ceramics, coatings, and energy storage. However, developing new routes for making nano-based materials is a challenging area for solid-state materials chemists. This book does just that by describing a novel method for preparing them. The authors have developed a novel low-temperature, self-propagating synthetic route to nano-oxides by the solution combustion and combustible precursor processes. This method provides the desired composition, structure, and