

1. Record Nr.	UNINA9910953235203321
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Titolo	Electron-phonon interactions in novel nanoelectronics / / Takashi Kato
Pubbl/distr/stampa	New York, : Nova Science, c2009
ISBN	1-61728-391-6
Edizione	[1st ed.]
Descrizione fisica	1 online resource (97 p.)
Soggetti	Electron-phonon interactions Nanoelectronics
Lingua di pubblicazione	Inglese
Formato	Materiale a stampa
Livello bibliografico	Monografia
Note generali	"Novinka."--Cover.
Nota di bibliografia	Includes bibliographical references (p. [79]-84) and index.
Nota di contenuto	<p>Intro -- ELECTRON-PHONONINTERACTIONS IN NOVELNANOELECTRONICS -- CONTENTS -- PREFACE -- INTRODUCTION -- THEORETICAL BACKGROUND -- A. Vibronic Interactions between the Twofold DegenerateFrontier Orbitals and the E2g Vibrational Modes InBenzene -- B. Vibronic Interactions between the Twofold DegenerateFrontier Orbitals and the A1g Vibrational Modes inBenzene -- C. Vibronic Interactions between the NondegenerateFrontier Orbitals and the Ag Vibrational Modes inPolyacenes -- D. Vibronic Interactions in Polyfluoroacenes, B, NSubstitutedPolyacenes, and Polycyanodienes -- ELECTRON-PHONON COUPLING CONSTANTSFOR THE CHARGED ELECTRONIC STATESOF POLYACENES, POLYFLUOROACENES, B,N-SUBSTITUTED POLYACENES, ANDPOLYCYANODIENES -- A. Twofold Degenerate Electronic States of TheMonocations of Monoanions of Benzene,Hexafluorobenzene, and Borazine -- B. Nondegenerate Electronic Systems of Polyacenes, Polyfluoroacenes, B, N-Substituted Polyacenes, andPolycyanodienes -- OPTIMIZED STRUCTURES -- A. Benzene and Polyacenes -- B. Polyfluoroacenes -- C. B, N-Substituted Polyacenes -- D. Polycyanodienes -- C. B, N-Substituted Polyacenes -- D. Polycyanodienes -- ELECTRON-PHONON COUPLING CONSTANTS -- A. The Monocations and Monoanions of Benzene andPolyacenes -- B. Polyfluoroacenes -- 1. Monocations -- 2. Monoanions -- C. B, N-Substituted Polyacenes -- 1. Monocations -- 2. Monoanions -- D. Polycyanodienes -- 1. Monocations -- 2. Monoanions -- E. Summary</p>

-- TOTAL ELECTRON-PHONON COUPLINGCONSTANTS -- A. Polyacenes and Polyfluoroacenes -- 1. Monocations -- 2. Monoanions -- B. B, N-Substituted Polyacenes -- C. Polycyanodienes -- 1. Monocations -- 2. Monoanions -- D. Summary -- THE LOGARITHMICALLY AVERAGED PHONONFREQUENCIES -- A. Polyacenes and Polyfluoroacenes -- 1. Monocations -- 2. Monoanions -- B. B, N-Substituted Polyacenes -- C. Polycyanodienes -- D. Summary -- CONCLUDING REMARKS. ACKNOWLEDGMENT -- REFERENCES -- INDEX.

Sommario/riassunto

In this book, the electron-phonon interactions in the charged molecular systems such as polyacenes, polyfluoroacenes, B, N-substituted polyacenes, and polycyanodienes are discussed. They estimated the electron-phonon coupling constants and the frequencies of the vibronic active modes playing an essential role in the electron-phonon interactions in order to discuss how CH-CF, CC-BN, and CC-CN substitutions are closely related to the essential characteristics of the electron-phonon interactions in these molecules by comparing the calculated results for charged polyacenes with those for charged B, N-substituted polyacenes and polycyanodienes, respectively. The C-C stretching modes around 1500 cm⁻¹ strongly couple to the highest occupied molecular orbitals (HOMO), and the lowest frequency modes and the C-C stretching modes around 1500 cm⁻¹ strongly couple to the lowest unoccupied molecular orbitals (LUMO) in polyacenes. The C-C stretching modes around 1500 cm⁻¹ strongly couple to the HOMO and LUMO in polyfluoroacenes. The B-N stretching modes around 1500 cm⁻¹ strongly couple to the HOMO and LUMO in B, N-substituted polyacenes. The C-C and C-N stretching modes around 1500 cm⁻¹ strongly couple to the HOMO and LUMO in polycyanodienes. The total electron-phonon coupling constants for the monocations () and monoanions () decrease with an increase in molecular size in polyacenes, polyfluoroacenes, B, N-substituted polyacenes, and polycyanodienes. In general, we can expect that monocations and monoanions, in which number of carriers per atom is larger, affords larger value. The CH-CF, CC-BN, and CC-CN atomic substitutions are effective way to seek for larger values, and the CH-CF and CC-CN atomic substitutions are effective way to seek for larger values in polyacenes. The logarithmically averaged phonon frequencies () which measure the frequencies of the vibronic active modes playing an essential role in the electron-phonon interactions for the monocations () and monoanions () in polyacenes, polyfluoroacenes, B, N-substituted polyacenes, and polycyanodienes are investigated. The values decrease with an increase in molecular size in polyacenes, polyfluoroacenes, and polycyanodienes, and the values decrease with an increase in molecular size in polyacenes, polyfluoroacenes, B, N-substituted polyacenes, and polycyanodienes. The authors can expect that in the hydrocarbon molecular systems, the values would basically decrease by substituting hydrogen atoms by heavier atoms. This can be understood from the fact that the frequencies of all vibronic active modes in polyacenes downshift by H-F substitution. However, considering that the value for the LUMO rather localized on carbon atoms in large sized polyfluoroacenes becomes larger by H-F substitution, the authors can expect that the value for a molecular orbital localized on carbon atoms has a possibility to increase by substituting hydrogen atoms by heavier atoms if the phase patterns of the molecular orbital do not significantly change by such atomic substitution. Therefore, the detailed properties of the vibrational modes and the electronic structures as well as the molecular weights are closely related to the frequencies of the vibronic active modes playing an important role in the electron-phonon interactions in the monoanions of polyfluoroacenes.

2. Record Nr.	UNINA9910892973603321
Titolo	Revista mexicana de investigacion educativa
Pubbl/distr/stampa	[Mexico, D.F.], : Consejo Mexicano de Investigacion Educativa, ©1996-
ISSN	2594-2271
Descrizione fisica	1 online resource (volumes)
Disciplina	370.952
Soggetti	Education - Mexico Education - Latin America Education Education (General) Latin America Mexico periodicals. Periodicals. Periodiques. Latin America Mexico
Lingua di pubblicazione	Spagnolo
Formato	Materiale a stampa
Livello bibliografico	Periodico
Note generali	Refereed/Peer-reviewed Title from issue contents page (viewed July 6, 2004). Some issues also have distinctive titles.