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Nota di contenuto	Chemical Synthesis Using Supercritical Fluids; Preface; Contents; Contents; List of Contributors; 1 Introduction; 1.1 Supercritical Fluids as Media for Chemical Reactions; 1.1.1 What is a Supercritical Fluid (SCF)?; 1.1.2 Practical Aspects of Reactions in Supercritical Fluids (SFRs); 1.1.3 Motivation for Use of SCFs in Modern Chemical Synthesis; 1.1.4 A Brief History of Chemical Synthesis in SCFs; 1.1.4.1 Discovery of SCFs and their Use as Solvents; 1.1.4.2 Early Examples of Chemical Reactions in SCFs; 1.1.4.3 Industrial Use of SCFs as Reaction Media; References 1.2 Phase Behavior and Solubility1.2.1 Basic Physical Properties of Supercritical Fluids; 1.2.2 Phase Behavior in High Pressure Systems; 1.2.2.1 Types of Binary Phase Diagrams; 1.2.2.2 Asymmetric Binary Mixtures; 1.2.3 Factors Affecting Solubility in Supercritical Fluids; 1.2.3.1 SCF Solvent; 1.2.3.2 Chemical Functionality of the Solute; 1.2.3.3 Temperature and Pressure Effects; References; 1.3 Physical Properties as Related to Chemical Reactions; 1.3.1 Behavior of Diffusion Coefficients; 1.3.2 Diffusional Effects on Reactions; 1.3.3 Transition-

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 2.1.2.6 Stirring and Mixing; 2.1.2.7 Optical Windows; 2.1 Pressure Systems; 2.1.3.1 Single-batch High-pressure F View Cells; 2.1.3.3 Systems for Continuous Processing 2.1.4 SummaryReferences; 2.2 Extraction and Related S Techniques; 2.2.1 General Aspects of Supercritical Fluic Separating Agents; 2.2.2 Extraction from Solids; 2.2.2.1 Design; 2.2.2.2 Process Parameters; 2.2.2.3 Modeling tf 2.2.2.4 Solids in Multiple Stages and Countercurrent Op 2.2.2.5 Continuous Extraction of Contaminated Soil with Water; 2.2.3 Countercurrent Multistage Extraction; 2.2.3 Process Design; 2.2.3.2 Phase Equilibria; 2.2.3.3 Separ- with Respect to Theoretical Stages 2.2.3.4 Multicomponent Process Simulation2.2.3.5 Dete the Height (Length) of a Theoretical Stage; 2.2.3.6 Deter Column Diameter; 2.2.4 Chromatographic Separation wi Fluids; 2.2.4.1 Design of SFC Apparatus; 2.2.4.2 Method of Chromatography; 2.2.5 Conclusion; References; 2.3 F Crystallization Techniques; 2.3.1 Introduction; 2.3.2 The and Phase Equilibria; 2.3.2.1 CSS, PGSS and RESS; 2. Process Basics and Reference Schemes; 2.3.3.1 Crystal Supercritical Solution (CSS) 2.3.3.2 Formation of Particles from Gas Saturated Solution 	ds as Mass Basic Process he Extraction; beration in SFE; a Supercritical 3.1 Basic ration Analysis for Analysis for Scale-up Precipitation and for Scale-up Precipitation and for Scale-up Precipitation and for Scale-up Precipitation and for Scale-up Precipitation and for Scale-up Precipitation and for Scale-up	
Sommario/riassunto For 'better solutions' - this practical guide describes how advantage of supercritical fluids in chemical synthesis. V in extractions and materials processing, supercritical flui becoming increasingly popular as media for modern che syntheses. Historically, the application of compressed garestricted mainly to the production of bulk chemicals. In the decade, however, research has turned to exploiting the properties of supercritical fluids for the synthesis of fine of specialized materials. Now that the necessary equipment	Well-established ids are emical ases has been the last unique chemicals and	