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Nota di contenuto	Chemical Synthesis Using Supercritical Fluids; Preface; Contents; Contents; List of Contributors; 1 Introduction; 1.1 Supercritical Fluids as Media for Chemical Reactions; 1.1.1 What is a Supercritical Fluid (SCF)?; 1.1.2 Practical Aspects of Reactions in Supercritical Fluids (SFRs); 1.1.3 Motivation for Use of SCFs in Modern Chemical Synthesis; 1.1.4 A Brief History of Chemical Synthesis in SCFs; 1.1.4.1 Discovery of SCFs and their Use as Solvents; 1.1.4.2 Early Examples of Chemical Reactions in SCFs; 1.1.4.3 Industrial Use of SCFs as Reaction Media; References 1.2 Phase Behavior and Solubility1.2.1 Basic Physical Properties of Supercritical Fluids; 1.2.2 Phase Behavior in High Pressure Systems; 1.2.2.1 Types of Binary Phase Diagrams; 1.2.2.2 Asymmetric Binary Mixtures; 1.2.3 Factors Affecting Solubility in Supercritical Fluids; 1.2.3.1 SCF Solvent; 1.2.3.2 Chemical Functionality of the Solute; 1.2.3.3 Temperature and Pressure Effects; References; 1.3 Physical Properties as Related to Chemical Reactions; 1.3.1 Behavior of Diffusion Coefficients; 1.3.2 Diffusional Effects on Reactions; 1.3.3 Transition-

## state Theory Applied to SCFs

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2.3.3.2 Formation of Particles from Gas Saturated Solution (PGSS)

### Sommario/riassunto

For 'better solutions' - this practical guide describes how to take advantage of supercritical fluids in chemical synthesis. Well-established in extractions and materials processing, supercritical fluids are becoming increasingly popular as media for modern chemical syntheses. Historically, the application of compressed gases has been restricted mainly to the production of bulk chemicals. In the last decade, however, research has turned to exploiting the unique properties of supercritical fluids for the synthesis of fine chemicals and specialized materials. Now that the necessary equipment is