Record Nr. UNINA9910830504203321 Chemical dynamics [[electronic resource]]: papers in honor of Henry **Titolo** Eyring / / edited by Joseph O. Hirschfelder [and] Douglas Henderson Pubbl/distr/stampa New York, : Wiley-Interscience, [1971] **ISBN** 1-282-34744-6 9786612347443 0-470-14369-X 0-470-14406-8 Descrizione fisica 1 online resource (852 p.) Collana Advances in chemical physics; ; v. 21 Altri autori (Persone) EyringHenry <1901-1981.> HendersonDouglas <1934-> HirschfelderJoseph O. <1911-1990> Disciplina 541.3 541.305 541/.08 Soggetti **Biophysics Dynamics Kinematics** Quantum theory Lingua di pubblicazione Inglese **Formato** Materiale a stampa Livello bibliografico Monografia Note generali Description based upon print version of record. Nota di bibliografia Includes bibliographical references. CHEMICAL DYNAMICS; CONTENTS; Helium Molecular States; A Free-Nota di contenuto Electron Study of Common Boron Frameworks: Basis of Extended Hiickel Formalism; Ionization Potentials and Electron Afinities of Some Polyphenyl Molecules; A Linear Sum-over-Points Approach for Computing Electronic Energies: Application to He and H2 Using Conroy Diophantine Points; The Potential Energy Surface of the H3 System Using Floating Gaussian Orbituls; A Forecast for Theoretical Chemistry; Studies on Rates of Nonequilibrium Processes: The Current Stutus of Eyring's Rute Threory Permeabilities for Reactions of the Type H+ H2= H2+H Treated as a

Linear Encounter Using Variational and Distorted Wave

TechniquesCalculation of Statistical Complexions of Polyatomic

Molecules and Ions; Some Comments on the Theory of Photochemical Reactions; Collisional Transfer of Triplet Excitations Between Helium Atoms; The Mechanism of Electronic Energy Transfer Between Excited Mercury (3PI) Atoms and Gaseous Paraffins; Dynamics of Ion-Molecule Collisions; Activation Parameters and Deprotonation Kinetics of Intrarnolecularly Hydrogen Bonded Acids

The Bronsted a and the Primary Hydrogen Isotope Effect. A Test of the Marcus TheorySelfDifusion of Oxygen in Iota Ph.ase Praseodymium Oxide; Application of the Absolute Reuction-Rate Theory to Non-New tonian Flow; Electronic States of Solid Explosives and Their Probable Role in Detonations; Notes on the Two-Purticle Density Matrix in n-Electron Theory; Tn Frequency Functions as Energy Contours for Photon Absorbance in Condensed Systems; Covalency Efects in Octahedral 5f1 Complexes; The Calculated Heat of Adsorption of Water on Mercury, Silver, Gold, and Platinum

Intermolecular Potentials and Macroscopic Properties of Argon and Neon from Differential Collision Cross SectionsDistribution Function of Classical Fluids of Hard Spheres. I; Hard Spheres with Surfuce Adhesion: The Percus-Yevick Approximation and the Energy Equation; Inequalities for Critical Indices near Gas-Liquid Critical Point; Application of a Short-Range Ordered Model to Strong Electrolytes; Eyring's Theory of' Viscosity of Dense Media and Noneyuilibriunz, Statistical Mechanics; Sound Velocity and van der Waals Force in Liquids According to Signiyicant Structure Theory

Transient State Theory of Significant Liquid Structure Applied to WaterThe Gaseous Fraction in Liquid Metals; Application of the Significant Structures Theory to Plastic Crystals; On the Nature of Solutions of Organic Compounds in Fused Salts; Solid "Liquid-Crystalline" Films of Synthetic Polypeptides: A New State of Matter; The Thermal Stability of Collagen: Its Significance in Biology and Physiology; Kinetics of the Interactions of Formaldehyde, Acetaldehyde, and Acrolein with Rattail Tendon; Protein Conformations, "Rack" Mechanisms and Water

Biomolecular Conformation and Biological Activity

Sommario/riassunto

The Advances in Chemical Physics series provides the chemical physics and physical chemistry fields with a forum for critical, authoritative evaluations of advances in every area of the discipline. Filled with cutting-edge research reported in a cohesive manner not found elsewhere in the literature, each volume of the Advances in Chemical Physics series serves as the perfect supplement to any advanced graduate class devoted to the study of chemical physics.