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| Note generali | Description based upon print version of record. |
| Nota di bibliografia | Includes bibliographical references and index. |
| Nota di contenuto | Electrochemical Water Processing; Contents; Preface; Acknowledgements; Introduction; 1. Water Contaminants and Their Removal; 1.1 Introduction; 1.2 Technology, History, and Background; 1.3 Application Areas: Electrochemical Technology Water Processing; 2. Basic Electrochemical and Physical Principles; 2.1 Introduction; 2.2 Acidity and Alkalinity, pH; 2.3 Activity and Activity Coefficients; 2.4 Equilibrium and Dissociation Constants; 2.4.1 Degree or Percentage Dissociation; 2.5 Electrode, or Half Cell Potential; 2.6 Chemical Potential Definition; 2.7 Concentration Potential 2.8 Equivalent Conductance2.9 Free Energy and Equilibrium; 2.10 Dissociation Constants; 2.11 Ionic Conductance and Mobility; 2.12 Osmotic Pressure; 2.13 Diffusion (Flick's Law); 3. Systems Description: General Outlines of Basic Approaches; 3.1 Electrodialysis; 3.1.1 Performance Characteristics; 3.1.2 General Purpose Processor; 3.1.3 Additional Details for Appropriate Application - Desalinator for Small Boats; 3.2 pH Control: Analytic Development; 3.2.1 Introduction; 3.2.2 Some Technical Background; 3.2.3 Sample Processes for pH Control; |

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| | 3.2.4 Application Possibilities 3.2.4.1 Swimming Pool Water3.2.4.2 Cooling Towers; 3.2.4.3 Regeneration of Ion Exchange Resins; 3.2.5 Current and Electrical Energy Requirements; 3.2.6 Shielded (Limited Ion Access) Positive Electrode Operation; 3.2.6.1 Double Barrier; 3.2.6.2 Close Spacing; 3.2.6.3 Porous Barrier Design; 3.2.6.4 Etched Electrode Surfaces; 3.3 Biociding Technology; 3.3.1 Electrolytic Production of Free Halogens; 3.3.2 Chlorination Process Description; 3.3.3 Bromination Process Description; 3.4 Ion Exchange Resin Regeneration System; 3.4.1 General; 3.4.1.1 Present Regeneration Methods 3.4.1.2 Electrochemical Regeneration Methods. 3.4.1.2 Electrochemical Regeneration Methods. 3.4.1.2 Electrochemical Regeneration Methods. 3.5.5 Metals Reclamation; 3.5.1 Electrochemical Process for the Removal of Iron in Acid Baths; 3.5.2 Technical Approaches; 3.5.3 Technical Approaches; 3.5.4 Laboratory Feasibility & Data Study Suggestions; 3.5.5 Experimental Methods; 3.5.5.1 Approach B Tests; 3.5.2 Approach A; 3.5.6 Conclusions & Recommendations; 4. Mathematical Analysis & Modeling Electrodialysis Systems; 4.1 Electrodialysis: Descriptions and Definitions; 4.2 Basic Assumptions and Operating Parameters 4.2.1 Electrolytic Conductivity4.2.2 Solute Concentration & Electrical Conduction; 4.2.3 Electric Charge Equivalence; 4.2.4 Coulombic Efficiency; 4.2.5 Coefficients of Performance; 4.3 Parametric Analysis: Flow-Through Configuration; 4.3.1 Performance Analysis of Electro- dialytic Systems, Part I; 4.3.1.1 First Approximation; 4.3.1.2 Design Assumptions; 4.3.1.3 Equation Development; 4.3.1.4 Resistance of a Cell; 4.3.2 Further Definition of Terms; 4.3.2.1 Average Current Density; 4.3.2.2 Entrance & Exit Current Densities; 4.3.2.3 Water Flow Rate in Processed Chamber 4.3.24 Solute Concentration Along the Length of the Cell |
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| Sommario/riassunto | Even though most of the Earth's surface is covered with water, most of it is not directly usable for human consumption or applications. As the population increases and our general style of living standards increase, the importance useable water is becoming acute. This book addresses this issue with approaches to treating water sources that require removal of unwanted or dissolved substances. In particular, it covers various methods for removing dissolved ionic materials. There are numerous methods for accomplishing this end, and the book reviews most of them in some depth. |