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Nota di contenuto	Cover; Title Page; Copyright; Contents; Preface; Symbols and Abbreviations; Chapter 1 Introduction; 1.1 Equilibrium Properties of Complex Systems; 1.1.1 General Definitions; 1.1.2 Equilibrium in the Solutions of Complex Compounds; 1.1.3 Distribution of Complexes and Ligands in the Solution; References; Chapter 2 Equilibrium Electrode Potentials; 2.1 Electrodes of the First Kind; 2.2 Equilibria Involving Ions of the Intermediate Oxidation State; 2.3 Electrodes of the Second Kind; 2.4 Open-Circuit Potentials: Examples of Experimental Investigations; 2.4.1 System Cu/Cu(I),CN- 2.4.2 System Cu/Cu(II), -AlanineReferences; Chapter 3 Mass Transport; 3.1 Two Models of Linear Mass Transport; 3.2 Other Cases of Diffusional Mass Transport; 3.3 Mass Transport of Chemically Interacting Particles; 3.4 Concentration Profiles; 3.4.1 Concentration Profiles in Ideally Labile Systems; 4.2 Potential Transients; 3.4.2 Concentration Profiles in Systems of Limited Lability; References; Chapter 4 Peculiarities of Electrochemical Processes Involving Labile Complexes; 4.1 Steady-State Voltammograms; 4.3 Current Transients; References

Chapter 5 Quantitative Modeling of Quasi-Reversible Electrochemical Processes Involving Labile Complexes of Metals
 5.1 Kinetic Equations; 5.2 Employment of Voltammetric Data; 5.2.1 Tafel Plots Normalized with Respect to the Surface Concentration of EAC; 5.2.2 Analysis of LPS Current Maxima; 5.3 Techniques Based on the Control of the Intensity of Forced Convection; 5.3.1 Isosurface Concentration Voltammetry; 5.3.2 Determination of the Exchange Current Density from Polarization Resistance; 5.3.3 Electrochemical Impedance Spectroscopy (EIS) under Forced Convection Conditions; References
 Chapter 6 Determination of Mechanism of Electrochemical Processes Involving Metal Complexes
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 Chapter 7 Adsorption; 7.1 Thermodynamic Aspects; 7.2 Model Aspects; References;
 Chapter 8 Electrochemical Processes in Real Systems; 8.1 Experimental Details; 8.2 Cyanide Systems; 8.2.1 System Cu|Cu(I), CN⁻; 8.2.2 System Ag|Ag(I), CN⁻; 8.2.3 System Au|Au(I), CN⁻; 8.3 Ecological Systems Containing Hydroxy Acids; 8.3.1 Electroreduction of Cu(II) Complexes; 8.3.2 Electroreduction of Sn(II) Complexes; 8.3.3 Electroreduction of Zn(II) Complexes
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 Chapter 9 Electrochemical Deposition of Alloys; 9.1 Mass Transport during the Codeposition of Metals; 9.2 Codeposition of Cobalt and Tin; 9.3 Deposition of Brass Coatings; 9.4 Deposition of Bronze Coatings; 9.4.1 Surface Activity of Polyethers on Copper and Tin Substrates; 9.4.2 Effect of Halides. Formation of Surface Complexes; 9.4.3 Effect of Length of the Hydrocarbon Chain; 9.4.4 Codeposition of Copper and Tin; 9.4.5 Related Phenomena: Current Oscillations; 9.5 Codeposition of Cobalt and Molybdenum; References
 Chapter 10 Spontaneous Formation of Photosensitive Cuprous Oxide Layers

Sommario/riassunto

This book aims to sequentially cover all the major stages of electrochemical processes (mass transport, adsorption, charge transfer), with a special emphasis on their deep interrelation. Starting with general considerations on equilibria in solutions and at interfaces as well as on mass transport, the text acquaints readers with the theory and common experimental practice for studying electrochemical reactions of metals complexes. The core part of the book deals with all important aspects of electroplating, including a systematic discussion of co-deposition of metals and formation of alloys.
