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Autore	Martyshko P. S (Petr Sergeevich)
Titolo	Inverse problems of electromagnetic geophysical fields // P.S. Martyshko
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2. Record Nr.	UNINA9910825597903321
Autore	Sarge Stefan Mathias
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Lingua di pubblicazione	Inglese
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Nota di contenuto	Calorimetry: Fundamentals, Instrumentation and Applications; Contents; Preface; List of Quantities and Units; Introduction: Calorimetry: Definition, Application Fields and Units; Definition of Calorimetry; Application Fields for Calorimetry; First Example from Life Sciences; Second Example from Material Science; Third Example from Legal Metrology; Units; Further Reading; References; Part One: Fundamentals of Calorimetry; 1 Methods of Calorimetry; 1.1 Compensation of the Thermal Effect; 1.1.1 Compensation by a Phase Transition; 1.1.2 Compensation by Electric Effects 1.2 Measurement of Temperature Differences 1.2.1 Measurement of Time-Dependent Temperature Differences; 1.2.2 Measurement of Local Temperature Differences; 1.2.2.1 First Example: Flow Calorimeter; 1.2.2.2 Second Example: Heat Flow Rate Calorimeter; 1.3 Summary of Measuring Principles; References; 2 Measuring Instruments; 2.1 Measurement of Amount of Substance; 2.1.1 Weighing; 2.1.2 Volume Measurement; 2.1.3 Pressure Measurement; 2.1.4 Flow Measurement; 2.2 Measurement of Electric Quantities; 2.3 Measurement of

Temperatures; 2.3.1 Thermometers; 2.3.1.1 Liquid-in-Glass Thermometers  
 2.3.1.2 Gas Thermometers 2.3.1.3 Vapor Pressure Thermometers;  
 2.3.1.4 Resistance Thermometers; 2.3.1.5 Semiconductors; 2.3.1.6  
 Pyrometers; 2.3.2 Thermocouples; 2.4 Chemical Composition;  
 References; 3 Fundamentals of Thermodynamics; 3.1 States and  
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 Forms of Energy, Fundamental Form, and Thermodynamic Potential  
 Function; 3.1.2.1 Fundamental Form; 3.1.2.2 Thermodynamic Potential  
 Function; 3.1.3 Equilibrium; 3.1.4 Reversible and Irreversible Processes;  
 3.1.5 The Laws of Thermodynamics; 3.1.5.1 The Zeroth Law; 3.1.5.2  
 The First Law  
 3.1.5.3 The Second Law 3.1.5.4 The Third Law; 3.1.6 Measurement of  
 Thermodynamic State Functions; 3.2 Phases and Phase Transitions;  
 3.2.1 Multiphase Systems; 3.2.2 Phase Transitions; 3.2.3 Gibbs Phase  
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 during Heat Exchange; 4.6 Conclusions Concerning Calorimetry;  
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 5.3 The Adiabatic Condition 5.4 The Scanning Condition; Reference; 6  
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 6.1.2 Second Example: Biological System; 6.1.3 Third Example: First-  
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 Calorimeters; 6.2.3 Differential Scanning Calorimeters; 6.3  
 Reconstruction of the True Sample Heat Flow Rate from the Measured  
 Function; 6.3.1 Reconstruction of the Temperature Field for Negative  
 Times  
 6.3.2 The Convolution Integral and Its Validity

## Sommario/riassunto

Clearly divided into three parts, this practical book begins by dealing  
 with all fundamental aspects of calorimetry. The second part looks at  
 the equipment used and new developments. The third and final section  
 provides measurement guidelines in order to obtain the best results.  
 The result is optimized knowledge for users of this technique,  
 supplemented with practical tips and tricks.