1. Record Nr. UNINA9910821293003321 Autore Levitin Valim Titolo Interatomic bonding in solids: fundamentals, simulation, applications / / Valim Levitin Pubbl/distr/stampa Weinheim an der Bergstrasse, Germany:,: Wiley-VCH,, 2014 ©2014 **ISBN** 3-527-67155-2 3-527-67158-7 3-527-67157-9 Descrizione fisica 1 online resource (322 p.) Disciplina 541.224 Soggetti Chemical bonds Density functionals - Computer simulation Materials science - Computer simulation Lingua di pubblicazione Inglese **Formato** Materiale a stampa Livello bibliografico Monografia Note generali Description based upon print version of record. Includes bibliographical references and index. Nota di bibliografia Nota di contenuto Cover: Title Page: Contents: Preface: 1 Introduction: 2 From Classical Bodies to Microscopic Particles; 2.1 Concepts of Quantum Physics; 2.2 Wave Motion; 2.3 Wave Function; 2.4 The Schrodinger Wave Equation; 2.5 An Electron in a Square Well: One-Dimensional Case; 2.6 Electron in a Potential Rectangular Box: k-Space; 3 Electrons in Atoms; 3.1 Atomic Units; 3.2 One-Electron Atom: Quantum Numbers; 3.3 Multi-Electron Atoms: 3.4 The Hartree Theory: 3.5 Results of the Hartree Theory: 3.6 The Hartree-Fock Approximation; 3.7 Multi-Electron Atoms in the Mendeleev Periodic Table: 3.8 Diatomic Molecules 4 The Crystal Lattice4.1 Close-Packed Structures; 4.2 Some Examples of Crystal Structures; 4.3 The Wigner-Seitz Cell; 4.4 Reciprocal Lattice; 4.5 The Brillouin Zone; 5 Homogeneous Electron Gas and Simple Metals: 5.1 Gas of Free Electrons: 5.2 Parameters of the Free-Electron Gas: 5.3 Notions Related to the Electron Gas; 5.4 Bulk Modulus; 5.5 Energy of Electrons; 5.6 Exchange Energy and Correlation Energy; 5.7

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Sommario/riassunto

The connection between a quantum behavior of the structure elements of a substance and the parameters that determine the macroscopic behavior of materials has a major influence on the properties exhibited by different solids. Although quantum theory and engineering should complement each other, this is not always the case. This book aims to demonstrate how the properties of materials can be derived and predicted proceeding from the features of their structural elements, generally electrons. In a sense, electronic structure forms the glue holding solids as whole, and it is centr