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Nota di contenuto	Intelligent Nanomaterials: Processes, Properties, and Applications; Contents; Preface; PART I Inorganic Materials; 1. Synthesis, Characterization, and Self-assembly of Colloidal Quantum Dots; 1.1 Introduction; 1.2 Size-dependent Optical Properties of Quantum Dots; 1.2.1 Band Gap Energies; 1.2.2 Absorption Spectra; 1.3 Procedures for Synthesis of Colloidal Quantum Dots; 1.3.1 Synthesis of Quantum Dots in Reverse Micelles; 1.3.2 Synthesis of Quantum Dots in Aqueous Media; 1.3.3 Hot-matrix Synthesis of Quantum Dots; 1.4 Types of Semiconductor Quantum Dots; 1.4.1 Binary Quantum Dots 1.4.2 Alloyed Quantum Dots: 1.4.3 Core/shell Quantum Dots: ""Type-I""; 1.4.4 Core/shell Quantum Dots: ""Type-II""; 1.4.5 Quantum Dot/quantum Well Nanocrystals; 1.4.6 Transition-element-doped Quantum Dots; 1.5 Surface Functionalization of Quantum Dots; 1.5.1 Self-assembly of Colloidal Quantum Dots; 1.6 Conclusions; References; 2. One-dimensional Semiconducting Metal Oxides: Synthesis, Characterization and Gas Sensors Application; 2.1 Introduction; 2.2

1.

	Synthesis of 1-D Metal Oxide; 2.2.1 Vapor Phase Growth; 2.2.2 Vapor- liquid-solid Mechanism; 2.2.3 Vapor Solid Mechanism; 2.3 Solution Phase Growth 2.3.1 Template Assisted Synthesis2.3.2 Template Free Synthesis; 2.4 Gas Sensor Applications; 2.4.1 SnO2 NWs Based Gas Sensors; 2.4.2 WO3 NWs Based Gas Sensors; 2.4.3 ZnO NWs Based Gas Sensors; 2.4.4 TiO2 NWs Based Gas Sensor; 2.4.5 CuO NWs Based Gas Sensors; 2.4.6 In2O3 NWs Based Gas Sensor; 2.5 Conclusions; Acknowledgement; References; 3. Rare-earth Based Insulating Nanocrystals: Improved Luminescent Nanophosphors for Plasma Display Panels; 3.1 What is Plasma Display Panel; 3.3 Working of Plasma Display Panel 3.3.1 Advantages of Plasma Display Panel 3.3.1 Advantages of Plasma Display Panel; 3.4.1 Blue Nanophosphors; 3.5 Synthesis of BAM:Eu2+ Nanophosphors by Sol-gel Method; 3.5.1 Chemicals Used; 3.5.2 Methodology; 3.5.3 Characterization of Prepared Nanophosphors by Solution Combustion Method; 3.7.1 Chemicals Used; 3.7.2 Methodology; 3.7.3 Characterization of Prepared Nanophosphors by Solution Combustion Method; 3.7.1 Chemicals Used; 3.7.2 Methodology; 3.7.3 Characterization of Prepared Nanophosphors; 3.7.4 Results and Discussion 3.8 Green Nanophosphors 3.8.1 Yttrium Aluminum Garnet Y3Al5O12: Tb3+ (YAG:Tb3+) Nanophosphors is Sol-gel Method; 3.8.2 Synthesis of Y3Al5O12:Tb3+ (YAG:Tb3+) Nanophosphors; 3.8.6 Results and Discussion 3.8.4 Methodology; 3.8.5 Characterization of Prepared Y3Al5O12:Tb3+ (YAG:Tb3+) Nanophosphors; 3.8.6 Results and Discussion; 3.9 Terbium Doped Yttrium Ortho-borate (YBO3:Tb3+) Nanophosphors; 3.9.1 Synthesis of Terbium Doped Yttrium Ottho-borate (YBO3:Tb3+) Nanophosphors; 3.9.2 Chemicals Used; 3.9.3 Methodology; 3.9.4 Characterizations Used; 3.9.5 Result and Discussion 3.10 Red Nanophosphors: Yttrium Aluminum Garnet Y3AlO12:Eu3+ (YAG:Eu3+) Nanophosphors; Yttrium Aluminum Garnet Y3AlO12:Eu3+ (YAG:Eu3+) Nanophosphors; 3.9.5 Result and Discussion 3.10 Red Nanophosphors; Yttrium Aluminum Garnet Y3AlO12:Eu3+
Sommario/riassunto	Intelligent Nanomaterials comprehensively provides up-to-date material of this fascinating field. The last three decades have seen extraordinary advances in the generation of new materials based on both fundamental elements and composites, driven by advances in synthetic chemistry and often drawing inspiration from nature. The concept of an intelligent material envisions additional functionality built into the molecular structure, such that a desirable response occurs under defined conditions. Divided into 4 parts: Inorganic Materials; Organic Materials; Composite Materials; and