

1. Record Nr.	UNINA9910462707003321
Autore	Hobson J. A (John Atkinson), <1858-1940., >
Titolo	The conditions of industrial peace // by J.A. Hobson
Pubbl/distr/stampa	Abingdon, Oxon ; ; New York, N.Y. : , : Routledge, , 2012
ISBN	0-203-11846-4 1-283-84503-2 1-136-30999-3
Descrizione fisica	1 online resource (125 p.)
Collana	Routledge revivals
Disciplina	331.1
Soggetti	Arbitration, Industrial Industrial organization Industrial policy Wages Electronic books.
Lingua di pubblicazione	Inglese
Formato	Materiale a stampa
Livello bibliografico	Monografia
Note generali	First published in 1927 by George Allen & Unwin Ltd.
Nota di bibliografia	Includes bibliographical references.
Nota di contenuto	Cover; Title Page; Copyright Page; THE CONDITIONS OF INDUSTRIAL PEACE; PREFACE; Table of Contents; I. THE CAUSES OF INDUSTRIAL CONFLICTS; II. FINDING A FAIR WAGE; III. THE POOL OF WEALTH; IV. COSTS AND SURPLUS; V. A PEACE POLICY FOR INDUSTRY
Sommario/riassunto	First published in 1927, Hobson's treatise on industrial conflict analyses the nature and causes of industrial disputes with the aim of finding an equitable means of settling them. Assessing the notion of a fair wage within the context of the pool of wealth, Hobson sets about creating a peace policy for industry. Set against the backdrop of economic downturn and struggle in the interwar years, in the years preceding the great depression, this is work of social, historical and economic interest.

2. Record Nr.	UNINA9910968446403321
Autore	Wu Chih <1936->
Titolo	Thermodynamics and heat powered cycles : a cognitive engineering approach // Chih Wu
Pubbl/distr/stampa	New York, : Nova Science Publishers, c2007
ISBN	1-60692-626-8
Edizione	[1st ed.]
Descrizione fisica	1 online resource (677 p.)
Disciplina	621.402/1
Soggetti	Thermodynamics - Data processing Heat engineering - Data processing
Lingua di pubblicazione	Inglese
Formato	Materiale a stampa
Livello bibliografico	Monografia
Note generali	Description based upon print version of record.
Nota di bibliografia	Includes bibliographical references (p. [651]) and index.
Nota di contenuto	<p> ""THERMODYNAMICS AND HEATPOWERED CYCLES:A COGNITIVEENGINEERING APPROACH""; ""NOTICE TO THE READER""; ""CONTENTS""; ""PREFACE""; ""ACKNOWLEDGEMENTS""; ""BASIC CONCEPTS""; ""1.1. THERMODYNAMICS""; ""Homework 1.1. Thermodynamics""; ""1.2. BASIC LAWS""; ""Homework 1.2. Basic Laws""; ""1.3. WHY STUDY THERMODYNAMICS?""; ""Homework 1.3. Why Study Thermodynamics?""; ""1.4. DIMENSIONS AND UNITS""; ""Example 1.4.1.""; ""Example 1.4.2.""; ""Homework 1.4. Dimensions and Units""; ""1.5. SYSTEMS""; ""Homework 1.5. Systems""; ""1.6. PROPERTIES OF A SYSTEM""; ""1.6.1. Volume (V)"" ""1.6.2. Density (I?) and Specific Volume (v)""""Example 1.6.1.""; ""1.6.3. Pressure (p)""; ""Example 1.6.3.1.""; ""Example 1.6.3.2.""; ""1.6.4. Temperature (T)""; ""Example 1.6.4.1.""; ""1.6.5. Energy (E)""; ""1.6.6. Enthalpy (H)""; ""1.6.7. Specific Heat (c, cp and cv)""; ""1.6.8. Ratio of the Specific Heats (k)""; ""1.6.9. Quality, Dryness and Moisture Content""; ""Example 1.6.9.1.""; ""1.6.10. Entropy (S)""; ""1.6.11. Point Function""; ""Homewok 1.6. Properties""; ""1.7. EQUILIBRIUM STATE""; ""Homework 1.7. Equilibrium State""; ""1.8. PROCESSES AND CYCLES"" ""Homework 1.8. Processes and Cycles""""1.9. CYCLEPAD""; ""1.9.1. Download""; ""1.9.2. Installation onto your own PC""; ""1.9.3. Contents""; ""1.9.4. Modes""; ""1.10. SUMMARY""; ""PROPERTIES OF THERMODYNAMIC SUBSTANCES""; ""2.1. THERMODYNAMIC SUBSTANCES""; ""Homework 2.1. Thermodynamic Substances""; ""2.2. </p>

PURE SUBSTANCES"; "Example 2.2.1."; "Example 2.2.2."; "Example 2.2.3."; "Example 2.2.4."; "Example 2.2.5."; "Example 2.2.6."; "Example 2.2.7."; "Example 2.2.8."; "Example 2.2.9."; "Example 2.2.10."; "Homework 2.2. Pure substances"; "2.3. IDEAL GASES" "Example 2.3.1." "Example 2.3.2."; "Example 2.3.3."; "Example 2.3.4."; "Example 2.3.5."; "Example 2.3.6."; "Example 2.3.7."; "Homework 2.3. Ideal gases"; "2.4. REAL GASES"; "Example 2.4.1."; "Homework 2.4. Real gases"; "2.5. INCOMPRESSIBLE SUBSTANCES"; "Example 2.5.1."; "Example 2.5.2."; "Example 2.5.3."; "Homework 2.5. Incompressible substances (Liquids and solids)"; "2.6. SUMMARY"; "FIRST LAW OF THERMODYNAMICS FOR CLOSED SYSTEMS"; "3.1. INTRODUCTION"; "Homework 3.1. Introduction"; "3.2. WORK"; "Example 3.2.1."; "Example 3.2.2."; "Example 3.2.3." "Homework 3.2. Work" "3.3. HEAT"; "Homework 3.3. Heat"; "3.4. FIRST LAW OF THERMODYNAMICS FOR A CLOSED SYSTEM"; "Example 3.4.2."; "Homework 3.4. First Law of Thermodynamics for a Closed System"; "3.5. FIRST LAW OF THERMODYNAMICS FOR A CLOSED SYSTEM APPLY TO CYCLES"; "Example 3.5.1."; "Homework 3.5. First Law of Thermodynamics for a Closed System Apply to Cycles"; "3.6. CLOSED SYSTEM FOR VARIOUS PROCESSES"; "3.6.1. Constant Volume (Isochoric or Isometric) Process"; "Homework 3.6.1. Constant Volume"; "3.6.2. Constant Pressure (Isobaric) Process" "Homework 3.6.2. Isobaric Process"

Sommario/riassunto

An engineering textbook that is the result of fourteen semesters of CyclePad usage and evaluation of a course designed to exploit the power of the software, and to chart a path that integrates the computer with education. It aims to give students a thorough grounding in both the theory and practice of thermodynamics.
