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Nota di contenuto	Front Cover; Advanced Thermodynamics for Engineers; Copyright Page; Contents; Preface; Structure; Symbols; Chapter 1. State of Equilibrium; 1.1 Equilibrium of a thermodynamic system; 1.2 Helmholtz energy (Helmholtz function); 1.3 Gibbs energy (Gibbs function); 1.4 The use and significance of the Helmholtz and Gibbs energies; 1.5 Concluding remarks; Problems; Chapter 2. Availability and Exergy; 2.1 Displacement work; 2.2 Availability; 2.3 Examples; 2.4 Available and non-available energy; 2.5 Irreversibility; 2.6 Graphical representation of available energy and irreversibility 2.7 Availability balance for a closed system2.8 Availability balance for an open system; 2.9 Exergy; 2.10 The variation of flow exergy for a perfect gas; 2.11 Concluding remarks; Problems; Chapter 3. Pinch Technology; 3.1 A heat transfer network without a pinch problem; 3.2 A heat transfer network with a pinch point; 3.3 Concluding remarks; Problems; Chapter 4. Rational Efficiency of a Powerplant; 4.1 The influence of fuel properties on thermal efficiency; 4.2 Rational efficiency; 4.3 Rankine cycle; 4.4 Examples; 4.5 Concluding remarks; Problems

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	Chapter 5. Efficiency of Heat Engines at Maximum Power5.1 Efficiency of an internally reversible heat engine when producing maximum power output; 5.2 Efficiency of combined cycle internally reversible heat engines when producing maximum power output; 5.3 Concluding remarks; Problems; Chapter 6. General Thermodynamic Relationships (single component systems, or systems of constant composition); 6.1 The Maxwell relationships; 6.2 Uses of the thermodynamic relationships; 6.3 Tds relationships; 6.4 Relationships between specific heat capacities; 6.5 The Clausius-Clapeyron equation 6.6 Concluding remarksProblems; Chapter 7. Equations of State; 7.1 Ideal gas law; 7.2 Van der Waals' equation of state; 7.3 Law of corresponding states; 7.4 Isotherms or isobars in the two-phase region; 7.5 Concluding remarks; Problems; Chapter 8. Liquefaction of Gases; 8.1 Liquefaction by cooling - method (i); 8.2 Liquefaction by expansion - method (ii); 8.3 The Joule-Thomson effect; 8.4 Linde liquefaction plant; 8.5 Inversion point on p-v-T surface for water; 8.6 Concluding remarks; Problems; Chapter 9. Thermodynamic Properties of Ideal Gases and Ideal Gas Mixtures of Constant Composition 9.1 Molecular weights9.2 State equation for ideal gases; 9.3 Tables of u (T) and h(T) against T; 9.4 Mixtures of ideal gases; 9.5 Entropy of mixtures; 9.6 Concluding remarks; Problems; Chapter 10. Thermodynamics of Combustion; 10.1 Simple chemistry; 10.2 Combustion of simple hydrocarbon fuels; 10.3 Heats of formation and heats of reaction; 10.4 Application of the energy equation to the combustion process - a macroscopic approach; 10.5 Combustion processes; 10.6 Examples; 10.7 Concluding remarks; Problems; Chapter 11. Chemistry of Combustion; 11.1 Bond energies and heats of formation 11.2 Energy of formation
Sommario/riassunto	Although the basic theories of thermodynamics are adequately covered by a number of existing texts, there is little literature that addresses more advanced topics. In this comprehensive work the author redresses this balance, drawing on his twenty-five years of experience of teaching thermodynamics at undergraduate and postgraduate level, to produce a definitive text to cover thoroughly, advanced syllabuses.The book introduces the basic concepts which apply over the whole range of new technologies, considering: a new approach to cycles, enabling their irreversibility to be taken into a