Record Nr. UNINA9910454232903321 Autore Hillert Mats <1924-> Titolo Phase equilibria, phase diagrams and phase transformations: their thermodynamic basis / / Mats Hillert [[electronic resource]] Cambridge:,: Cambridge University Press,, 2008 Pubbl/distr/stampa 1-107-17647-6 **ISBN** 1-283-32974-3 9786613329745 1-139-13425-6 0-511-81278-7 1-139-12921-X 1-139-13141-9 0-511-50406-3 0-511-50620-1 Edizione [Second edition.] Descrizione fisica 1 online resource (xiv, 510 pages) : digital, PDF file(s) Disciplina 541.363 Soggetti Phase diagrams Phase transformations (Statistical physics) Lingua di pubblicazione Inglese **Formato** Materiale a stampa Livello bibliografico Monografia Title from publisher's bibliographic system (viewed on 05 Oct 2015). Note generali Nota di bibliografia Includes bibliographical references and index. Nota di contenuto Basic concepts of thermodynamics -- Manipulation of thermodynamic quantities -- Systems with variable composition -- Practical handling of multicomponent systems -- Thermodynamics of processes --Stability -- Applications of molar Gibbs energy diagrams -- Phase equilibria and potential phase diagrams -- Molar phase diagrams --Projected and mixed phase diagrams -- Direction of phase boundaries -- Sharp and gradual phase transformations -- Transformations in closed systems -- Partitionless transformations -- Limit of stability and critical phenomena -- Interfaces -- Kinetics of transport processes --Methods of modelling -- Modelling of disorder -- Mathematical

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Computational tools allow material scientists to model and analyze

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increasingly complicated systems to appreciate material behavior. Accurate use and interpretation however, requires a strong understanding of the thermodynamic principles that underpin phase equilibrium, transformation and state. This fully revised and updated edition covers the fundamentals of thermodynamics, with a view to modern computer applications. The theoretical basis of chemical equilibria and chemical changes is covered with an emphasis on the properties of phase diagrams. Starting with the basic principles, discussion moves to systems involving multiple phases. New chapters cover irreversible thermodynamics, extremum principles, and the thermodynamics of surfaces and interfaces. Theoretical descriptions of equilibrium conditions, the state of systems at equilibrium and the changes as equilibrium is reached, are all demonstrated graphically. With illustrative examples - many computer calculated - and worked examples, this textbook is an valuable resource for advanced undergraduates and graduate students in materials science and engineering.