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Outlook; Acknowledgments; References; Appendix; SPIN ORBIT COUPLING METHODS AND APPLICATIONS TO CHEMISTRY; I. Introduction; II. Theory and Methods; 1. general remarks; 2. electron correlation and SOC; III. Applications; 1. Hydrides of transition metals; 2. SOC in light diatomic molecules; 3. SOC in U and UF; 4. SOC in polyatomic molecules; Summary; Acknowledgements; References; TRANSGRESSING THEORY BOUNDARIES: THE GENERALIZED DOUGLAS-KROLL TRANSFORMATION
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2. Two-component relativistic quantum chemistry; 2.1. Basic properties of Dirac 4-spinors; 2.2. Elimination techniques; 2.3. Transformation techniques; 3. The generalized Douglas-Kroll transformation; 3.1. General parametrization of unitary transformations; 3.2. Derivation of the standard Douglas-Kroll Hamiltonians; 3.3. DK transformation of the two-electron terms; 3.4. Implementation of the DK transformation; 4. Results; 4.1. One-electron systems; 4.2. Many-electron atoms; 5. Conclusion; References
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Sommario/riassunto

Relativistic effects, though minor in light atoms, increase rapidly in magnitude as the atomic number increases. For heavy atom species, it becomes necessary to discard the Schrodinger equation in favor of the Dirac equation. Construction of an effective many-body Hamiltonian that accurately accounts for both relativistic and electron correlation effects in many-electron systems is a challenge. It is only in the past 20-25 years that relativistic quantum chemistry has emerged as a field of research in its own right, and it seems certain that relativistic many-electron calculations of molecular
