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Titolo	The Formation of bonds to Group VIB (O, S, Se, Te, Po) elements . Part 2 [[electronic resource] /] / founding editor, J. J. Zuckerman; editor, A. P. Hagen
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Altri autori (Persone)	HagenA. P ZuckermanJ. J <1936-1987.> (Jerold J.)
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Nota di contenuto	Inorganic Reactions and Methods; Contents; How to use this Book; Preface to the Series; Editorial Consultants to the Series; Contributors to Volume 6; Formation of the Group VIB (O, S, Se, Te, Po)- Group IB (Cu, Ag, Au) or IIB (Zn, Cd, Hg) Metal Bond; Introduction; Formation of the Oxygen-Group IB or IIB Bond; From Dioxygen and Ozone; By Reaction with the Metal; By Addition to Low-Valent and Unsaturated Metal Complexes; By Insertion into Metal-Ligand Bonds; From Water and Alcohols; By Reaction with Elements; By Hydrolysis or Alcoholysis of Metal-Ligand Bonds From Hydrogen Peroxide and Organic PeroxidesBy Oxidation of the Metals and Their Complexes; By Oxidation of the Ligands Coordinated to the Metals; From Neutral Oxygen Donor Ligands [Ethers, Aldehydes, Ketones, Pyridine N-Oxides, Phosphine Oxides, Arsine Oxides and Dialkyl(aryl) Sulfoxides]; By Ligand Displacement Reactions with

Transition Metal Complexes; By Insertion into the Metal-Ligand Bonds;
 From Bidentate and Polydentate Oxygen Donor Ligands (from
 Polyethers and Crown Ethers, Macrocycles, 2,4-Pentanedione, etc.);
 From Oxides of the Main Group Elements; By Reaction with the Metals
 By Reaction with Complexes of the Metals By Insertion into Metal-Ligand
 Bonds; From OH-, OR-, O²/2-, O²-; By Ligand Substitution Reactions
 with Complexes of the Metals; By Oxidation of the Metals and Their
 Complexes; By Metal Atom and Related Reactions; Formation of the
 Bond Between Sulfur and a Group IB or IIB Element; From Sulfur; By
 Direct Reaction with Metals; By Reaction with Metal Complexes; From
 Hydrogen Sulfide, Hydrogen Polysulfides, and Thiols; From
 Thiocarbonyls, Thioethers, Organic Polysulfides, and Other Sulfur
 Donor Ligands
 From Organic Thio Acids and Other Thio Acids of Main Group
 Elements By Oxidation of the Metals or Their Complexes; By Ligand
 Replacement Reactions with Complexes of the Metals and by Sulfur
 Atom Abstraction; From Bidentate and Polydentate Sulfur Donor Atoms;
 By Sulfur Addition, Oxidation and Sulfur Abstraction Reactions; By
 Ligand Substitution Reactions; From Sulfur Containing Anions (S²⁻,
 S_x²⁻, [HS-], [RS-]); By Metal Atom and Related Reactions; Formation of
 the Bond Between Selenium, Tellurium, and Polonium and Group IB or
 IIB Elements; By Reactions with the Group IB and IIB Metals
 Formation of the Bond with Selenium Formation of the Bond with
 Tellurium; Electrolytic Reactions Between the Elements; By Reaction
 with Group IB or Group IIB Metal Compounds; Binary Compounds;
 Ternary Compounds; By Reactions of Binary Acids of Selenium and
 Tellurium and Their Derivatives with Metal Compounds; By Reaction of
 Oxides of Selenium and Tellurium with Metal Compounds; By Reactions
 of the Anions and Oxyanions of the Elements with Metal Compounds;
 From Donor Ligands Incorporating the Elements Selenium and
 Tellurium; By Reaction with the Metals; Chemically Driven Reactions
 Electrochemically Driven Reactions

Sommario/riassunto

Boasting numerous industrial applications, inorganic chemistry forms
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