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Titolo	Annual review of gerontology and geriatrics . Volume 13 Focus on kinship, aging, and social change [[electronic resource] /] / George L. Maddox, M. Powell Lawton, volume editors
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Altri autori (Persone)	LawtonM. Powell <1923-2001.> (Mortimer Powell) MaddoxGeorge L
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Demographic, and Social Perspectives; Part V: Social Change and the Future of Kin Relationships; Chapter 10 Demographic Change and Kin Relationships in Later Life

Chapter 11 Kinship and Family Policy in an Aging SocietyIndex; A; B; C; D; E; F; G; H; I; J; K; L; M; N; O; P; R; S; T; U; V; W

**Sommario/riassunto**

During the past few decades, the dramatic social changes with regard to our aging population and changes in the family unit have made both demographic and socioeconomic consequences, as well as an effect on matters of social policy. The prestigious editors, George L. Maddox and M. Powell Lawton, have assembled an impressive group of expert contributors whose chapters address topics from the latest theory and research findings to the changing balance of work and families, as well as patterns of kinship.

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Sangwal Keshra

Additives and crystallization processes [[electronic resource]] : from fundamentals to applications / Keshra Sangwal

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## Nota di contenuto

Additives and Crystallization Processes; Contents; Preface; 1 Complexes in Solutions; 1.1 Structure of Common Solvents; 1.2 Structure of Pure Aqueous Electrolyte Solutions; 1.2.1 Solvation of Electrolyte Ions in Solutions; 1.2.2 Concentrated and Saturated Electrolyte Solutions; 1.2.3 Formation of Aquo and Partially Aquo Complexes; 1.3 Structure of Aqueous Electrolyte Solutions Containing Additives; 1.4 Polyelectrolytes and Surfactants in Solutions; 1.5 Polydentate Ligands and Molecular Additives; 1.6 Crystal-Additive Interactions; References

2 Three-Dimensional Nucleation and Metastable Zone Width 2.1 Driving Force for Phase Transition; 2.2 Three-Dimensional Nucleation of Crystals; 2.2.1 Three-Dimensional Nucleation Rate; 2.2.2 Three-Dimensional Heterogeneous Nucleation; 2.3 Metastable Zone Width; 2.4 Nucleation and Transformation of Metastable Phases; 2.4.1 Crystallization of Metastable Phases; 2.4.2 Overall Crystallization; 2.5 Induction Period for Crystallization; 2.6 Effects of Additives; 2.6.1 Solubility; 2.6.2 Three-Dimensional Nucleation Rate; 2.6.3 Metastable Zone Width; References

3 Kinetics and Mechanism of Crystal Growth: An Overview 3.1 Crystal Growth as a Kinetic Process; 3.2 Types of Crystal-Medium Interfaces; 3.3 Roughening of Steps and Surfaces; 3.3.1 Thermodynamic Roughening and the Surface Entropy Factor; 3.3.2 Kinetic Roughening; 3.4 Growth Kinetics of Rough Faces; 3.5 Growth Kinetics of Perfect Smooth Faces; 3.6 Growth Kinetics of Imperfect Smooth Faces; 3.6.1 Surface Diffusion and Direct Integration Models; 3.6.2 Bulk Diffusion Models; 3.6.3 Growth by a Group of Cooperating Screw Dislocations; 3.6.4 Preferential Growth at Edge Dislocations

3.7 Effect of Foreign Substances on Growth Kinetics 3.7.1 Some General Considerations; 3.7.2 Growth Kinetics by Heterogeneous Two-Dimensional Nucleation; 3.8 Real Crystal Growth Mechanisms; 3.8.1 Structure of Interfacial Layer; 3.8.2 Sources of Growth Steps; 3.9 Techniques for Studying Growth Kinetics; References; 4 Effect of Impurities on Crystal Growth Kinetics; 4.1 Mobile and Immobile Impurities; 4.2 Surface Coverage and Adsorption Isotherms; 4.2.1 Adsorption Isotherms; 4.2.2 Changes in Surface Free Energy by Adsorption of Impurities; 4.3 Kinetic Models of Impurity Adsorption 4.3.1 Earlier Models 4.3.2 Velocity of Curved Steps; 4.3.3 Impurity Adsorption at Kinks in Steps: Kubota-Mullin Model; 4.3.4 Impurity Adsorption at Surface Terrace: Cabrera-Vermilyea Model; 4.3.5 Effectiveness Factor for Impurity Adsorption; 4.3.6 Adsorption of Two Competing Impurities; 4.4 Confrontation of Impurity Adsorption Mechanisms with Experimental Data; 4.5 Time-Dependent Impurity Adsorption; 4.6 Growth Kinetics in the Presence of Impurities; 4.6.1 Basic Kinetic Equations; 4.6.2 Time Dependence of Face Displacement 4.6.3 Dependence of Kinetic Coefficient for Step Motion on Impurity Concentration

## Sommario/riassunto

Crystal growth technology involves processes for the production of crystals essential for microelectronics, communication technologies, lasers and energy producing and energy saving technology. A deliberately added impurity is called an additive and in different industries these affect the process of crystal growth. Thus, understanding of interactions between additives and the crystallizing phases is important in different processes found in the lab, nature and in various industries. This book presents a generalized description of the mechanisms of action of additives during nucleation, grow

3. Record Nr.	UNINA9910782121203321
Autore	Costisor Otilia
Titolo	Metal mediated template synthesis of ligands [[electronic resource] /] / Otilia Costisor, Wolfgang Linert
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;	4.2.2. NO donor ligands	4.2.3.	SN donor ligands
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	4.5.2. P- As amine cage ligands		
	4.5.3. Thioamine cage ligands		

## Sommario/riassunto

This book surveys the relatively new area of the synthesis of organic ligands when metal ions act as a template. In the last fifty years this field has undergone an explosive development, marked by a great amount of literature. The material in the book has been arranged according to the type of chemical reaction involved. In this frame, the basic principles of metal template reactions and the shape of the molecules are considered. Designed to satisfy the demands of students, young researchers doing their PhDs, and those working in the field of coordination chemistry, the book details the role