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2.2.2. Component activities in the intermediate phases
 2.3. Standard Gibbs free energy: temperature diagrams for oxides - Ellingham-Richardson diagrams; 2.3.1. Stoichiometric oxides; 2.3.2. Unstoichiometric compounds; 2.3.3. Thermodynamic data for the reduction of oxides by a reducing gas; 2.4. Thermodynamic data for sulfides and chlorides; 2.4.1. Ellingham-Richardson diagram for sulfides; 2.4.2. Stability diagrams for the (M-O-S) systems; 2.4.3. Ellingham-Richardson diagram for chlorides; 2.4.4. Stability diagrams of M-O₂-Cl₂ systems
 2.5. Metal-carbon phase diagrams and the Ellingham-Richardson diagram for carbides
 2.6. Carbon and carbon oxide reactions; 2.6.1. Oxidation reactions; 2.6.2. Boudouard's reaction; 2.6.3. The different types of coal; 2.7. Bibliography; Chapter 3. Metal Solutions, Slags and Mattes; 3.1. Introduction; 3.2. Metal solutions; 3.2.1. Phase diagrams and activities of liquid alloys components; 3.2.2. Activities and solubilities of metalloids in metal solutions; 3.2.3. Solubility and precipitation of oxide and sulfide compounds in metals; 3.3. Mattes
 3.3.1. Structure and physical properties of sulfide melts (mattes)
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Sommario/riassunto

This book is dedicated to the processes of mineral transformation, recycling and reclamation of metals, for the purpose of turning metals and alloys into a liquid state ready for pouring. Even though "process metallurgy" is one of the oldest technologies implemented by man, technological innovation, with the development of processes that are both focused on product quality and economically and ecologically efficient, continues to be at the heart of these industries. This book explains the physico-chemical bases of transformations, vital to their understanding and control (optimization of
